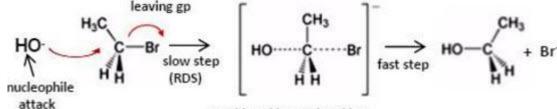


# End Semester/Reappear (Semester II) Examination July 2022

Su Su	bject: Pl bject Co	he: B. Pharm harmaceutical Organic Chemistry –I  bde: BP202T t No:
Section I		
1	Objecti	ive type questions. Answer all questions. $20 \times 1 = 20$
1.	i.	Primary alcohol is obtained by
	1.	a. Hydration of alkene c. dehydration of alkenes
		b. Dehydrogenation of alkenes d. None of above
	ii.	Alcohols reacts with PCl5 to give
	11.	a. Alkene b. Alkyl halide c. Acid d. None of these
	iii.	What is the correct IUPAC name of
	111.	CHO
		$NO_2$
		OCH <sub>3</sub>
		<ul> <li>a. 4-methoxy-2-nitrobenzaldehyde</li> <li>b. 4-methoxy-6-nitrobenzaldehyde</li> <li>c. 4-formyl-3-nitro anisole</li> <li>d. 2-formyl-5-methoxy nitrobenzene</li> </ul>
	iv.	Paraffin is a
	IV.	a. Alcohol b. Alkane c. Acid d. Ether
	v.	Primary alcohol is obtained by
	٧.	a. Hydration of alkene b. dehydration of alkenes
		b. Dehydrogenation of alkenes  d. None of above
	vi.	Which one of the following compounds is an isomer of CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OH?
	V 1.	a. CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> OH b. CH <sub>3</sub> CH(OH)CH <sub>3</sub> c. CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CHO d. CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>
	vii.	Which of the following will undergo an addition reaction with chlorine?
	VII.	a. CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub> b. CH <sub>3</sub> CH <sub>2</sub> CH=CHCH <sub>3</sub> c. C <sub>6</sub> H <sub>6</sub> d. CH <sub>3</sub> CH <sub>2</sub> COOH
	viii.	Dehydration of an alcohol leads to the formation of an
	, 111	a. Alkene b. Alkane c. Alkyne d. Alkyl halide
	ix.	Ethanol can be oxidized stepwise. What is the first stable intermediate product when ethanol
		is oxidized with a mild oxidation agent?
		a. CH <sub>3</sub> COOH b. CO <sub>2</sub> c. CH <sub>3</sub> CHO d. CH <sub>3</sub> CH <sub>2</sub> OH
	х.	According to anti Markovnikov's rule when propylene reacts with HBr in presence of
		peroxide product is
		a. n – propyl bromide b. Isopropyl bromide c. Butyl bromide d. Isobutyl bromide
	xi.	Which of the following compound would react most rapidly in an SN2 reaction?
		a. $CH_3 CH_2 I$ b. $CH_2 = CHI$ c. $(CH_3)_2 CHI$ d. $(CH_3)_3 CI$
	xii.	1, 2 dibromo ethane is formed by the reaction between
		a. $CH_2 = CH_2$ and $Br_2$ c. Acetylene with excess of HBr
		b. Ethylene with excess of HBr d. 1, 2 dichloro ethane with Br2
	xiii.	What is the IUPAC name of given structure $CH_3 - CH = CH - CH_2 - OH$
		a. $2 - buten - 1 - ol$ c. $2 - buten - 4 - ol$
		b. $2 - buten - 2 - ol$ d. $2 - butan - 4 - ol$

xiv. The reaction mechanism for the following reaction is



Bond breaking and making in transition state

- a. SN1 b. SN2
- c. E1
- d. E2

intermediate state.

- xv. SN1 mechanism proceeds
  - b. Carbonium ion
- c. Pentavalent transition
- d. Carbanion state

xvi. Cis-trans isomers are:

a. Free radical

- a. Diastereomers. isomers.
- b. Enantiomers
- c. Geometric isomers. d. Constitutional
- xvii. In the preparation of Grignard reagent from haloalkane, the metal \_\_\_\_\_ is used
  - a. Mg
- b. Zn
- c. Li
- d. K

- xviii. Diel's Alder reaction is
  - a. [2+2]-Cycloaddition reaction
- c. [4+2]-Cycloaddition reactions
- b. [4+4]-Cycloaddition reaction
- d. [6+2]-Cycloaddition reaction
- xix. Which of the following factors has no effect on the rate of SN1 reactions?
  - a. the concentration of the alkyl halide
- c. the concentration of the nucleophile
- b. the nature of the leaving group
- d. the value of the rate constant

- xx. Benzoic acid is used as
  - a. Preservative
- b. pH modifier
- c. Cosolvent
- d. None of the above

#### **Section II**

### 2. Short Answer type questions. Answer any five.

 $5 \times 7 = 35$ 

- a. Discuss basicity and effect of substitution on basicity of aliphatic amines
- b. Discuss effect of substitution on acidity of organic acids.
- c. Discuss structure and uses of
  - i. Acetic acid ii. Acetyl Salicylic acid
- d. Discuss the structures and uses of vanillin and cinnamaldehyde
- e. Give the qualitative tests for alcohols.
- f. Discuss the classification of ketones.
- g. Discuss factors affecting E1 and E2 reactions.

### **Section III**

## Long Answer type questions. Answer any two.

 $2 \times 10 = 20$ 

- 3. Explain the mechanism of Aldol condensation and Cannizaro reaction in details.
- 4. Discuss the procedure for IUPAC nomenclature for organic compounds
- 5. Write an note on structural isomerism on organic compounds.

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